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Lectures on
Dynamics of Combustion Waves
in Premixed Gases

Professor Paul Clavin
Aix-Marseille Université
ECM & CNRS (IRPHE)

Lecture II
Governing equations

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Lecture 2: **Governing equations**

(simplified form, see de Groot et Mazur (1962) or Williams (1985) for more details)

2-1. Conserved extensive quantities

2-2. Continuity

2-3. Fick's law. Diffusion equation

2-4. Conservation of momentum

2-5. Conservation of total energy

Thermal equation

Inviscid flows in reactive gases

Conservative forms

One-dimensional inviscid and compressible flow

2.6. Entropy production

II – 2) Conservation of mass: continuity equation

mass is a conserved scalar (classical mechanics)

$$\partial\rho/\partial t = -\nabla\cdot\mathbf{J} \quad \mathbf{J} \equiv \rho\mathbf{u} \quad \partial\rho/\partial t = -\nabla\cdot(\rho\mathbf{u})$$

material (convective) derivative $\frac{D}{Dt} \equiv \frac{\partial}{\partial t} + \mathbf{u}\cdot\nabla$

$$\frac{1}{\rho} \frac{D\rho}{Dt} = -\nabla\cdot\mathbf{u}$$

continuity equation

$$\frac{1}{v} \frac{Dv}{Dt} = \nabla\cdot\mathbf{u} \quad v \equiv 1/\rho$$

Lagrangian form of conservation equations

$$\partial(\rho a)/\partial t = -\nabla\cdot\mathbf{J}_a + \dot{\omega}_a$$

$$\rho Da/Dt = -\nabla\cdot\mathbf{J}'_a + \dot{\omega}_a$$

$$\mathbf{J}_a \equiv \rho a \mathbf{u} + \mathbf{J}'_a$$

convection flux

conserved scalar:

$$\partial(\rho a)/\partial t = -\nabla\cdot\mathbf{J}_a$$

$$\rho Da/Dt = -\nabla\cdot\mathbf{J}'_a$$

diffusion flux



(definition of the diffusion flux in the equation for energy is slightly different) see slide 11

II – 3) Fick's law. Diffusion equation

mass fraction $Y_i = \rho_i/\rho$ $\sum_i Y_i = 1$

inert mixture mass fraction of species is a conserved scalar

$$\rho D Y_i / Dt = -\nabla \cdot \mathbf{J}'_i \quad \sum_i \mathbf{J}'_i = 0$$

Kinetic theory of gas (binary diffusion in an abundant species)

Fick's law :

$$\mathbf{J}'_i = -\rho D_i \nabla Y_i$$

$$D_i > 0$$

$$\rho D Y_i / Dt = \nabla \cdot [\rho D_i \nabla Y_i]$$

diffusion equation

$$\rho D_i \approx \text{cst.} \quad \mathbf{u} = 0$$

$$\partial Y_i / \partial t = D_i \Delta Y_i$$

archetype of irreversible phenomenon

(random walk)

Diffusive damping. Dissipative phenomenon

$$\partial Y / \partial t = D \Delta Y \quad D > 0 \quad [D] = (\text{length})^2 / \text{time}$$

Fourier analysis

$$Y(\mathbf{r}, t) = \sum_{\mathbf{k}} \tilde{Y}_{\mathbf{k}}(t) e^{i\mathbf{k} \cdot \mathbf{r}} \quad k = |\mathbf{k}|$$

$$d\tilde{Y}_{\mathbf{k}}(t)/dt = -(Dk^2)\tilde{Y}_{\mathbf{k}}(t) \quad \tilde{Y}_{\mathbf{k}}(t) = \tilde{Y}_{\mathbf{k}}(0)e^{-Dk^2t}$$



Fourier 1824

Green function Self-similar solution

$$\partial G / \partial t = D \Delta G$$

$$t = 0 : G(\mathbf{r}, 0) = \delta(\mathbf{r}) \quad G(\mathbf{r}, t) = \frac{1}{(4\pi Dt)^{3/2}} e^{-r^2/4Dt} \quad r = |\mathbf{r}| \quad \iiint G(\mathbf{r}, t) d^3 \mathbf{r} = 1$$

probability distribution of the test particle

number density

$$\partial n / \partial t = D \Delta n$$

$$n(\mathbf{r}, t) = NG(\mathbf{r}, t)$$

$$n(\mathbf{r}, t) = \iiint n(\mathbf{r}', 0) G(\mathbf{r} - \mathbf{r}', t) d^3 \mathbf{r}'$$

II – 4) Conservation of momentum

Momentum is a conserved vector (isolated system)

$$\rho D\mathbf{u}/Dt = -\nabla \cdot \underline{\underline{\Pi}} - \rho g \mathbf{e}_z,$$

surface force (stress tensor) $\underline{\underline{\Pi}} = p \underline{\underline{\mathbf{I}}} + \underline{\underline{\pi}}$ gravity (body force)

thermodynamic pressure (isotropic)

Viscous stress tensor

$$\underline{\underline{\pi}} \equiv -2\eta(\underline{\underline{\nabla}}\mathbf{u})^{(s)} - \underline{\underline{\mathbf{I}}}(\xi - 2\eta/3)\nabla \cdot \mathbf{u}$$

Navier Stokes equations

$$\rho D\mathbf{u}/Dt = -\nabla(p + \rho g z) + \eta \Delta \mathbf{u} + (\xi + \eta/3)\nabla(\nabla \cdot \mathbf{u})$$

Viscous shear diffusivity

$$D_{vis} = \eta/\rho$$

Euler equations

$$\rho D\mathbf{u}/Dt = -\nabla p$$

non dissipative equations

II – 5) Conservation of **total** energy

internal energy, $e_{int} = e_T + e_{chem}$
 (Additive in a gas when interactions are neglected)
 thermal energy, (kinetic + rotational & vibrational energy)
 chemical energy (chemical bonds),

total energy $e_{tot} = |\mathbf{u}|^2/2 + e_T + e_{chem} + \dots$

total energy is a conserved scalar

$$\partial(\rho e_{tot})/\partial t = -\nabla \cdot \mathbf{J}_{e_{tot}} \quad \rho D e_{tot}/Dt = -\nabla \cdot \mathbf{J}'_{e_{tot}} \quad \mathbf{J}'_{e_{tot}} \equiv \mathbf{J}_{e_{tot}} - \rho e_{tot} \mathbf{u}$$

Inviscid and inert flows

(first law of thermodynamics)

Euler equation $\Rightarrow \frac{1}{2} \rho \frac{D}{Dt} |\mathbf{u}|^2 = -\mathbf{u} \cdot \nabla p = -\nabla \cdot (p\mathbf{u}) + p \nabla \cdot \mathbf{u}$

$\mathbf{J}'_q \equiv \mathbf{J}'_{e_{tot}} - p\mathbf{u}$
 heat flux,

$$\rho D(e_T + e_{chem})/Dt = -\nabla \cdot \mathbf{J}'_q - p \nabla \cdot \mathbf{u} \quad p \nabla \cdot \mathbf{u} = \rho p (D\rho^{-1}/Dt)$$

heat work done

Fourier law

(simplest form of heat flux)

Fourier equation

$$\mathbf{J}'_q = -\lambda \nabla T,$$

thermal conductivity

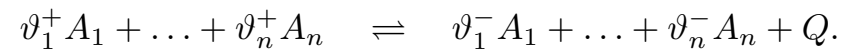
inert material $e_{chem} = \text{cst.}$
 no flow $\mathbf{u} = 0$

$$\partial T / \partial t = D_T \Delta T$$

$\delta e_T = c_V \delta T$ thermal diffusivity $D_T \equiv \lambda / \rho c_V$ $[D_T] = (\text{length})^2 / \text{time}$

Reactive flows

elementary reaction



reaction rate
nb/(volume × time)

$$\dot{W}^{(j)} \equiv (J_+^{(j)} - J_-^{(j)}) \quad \text{and} \quad \vartheta_i^{(j)} \equiv (\vartheta_i^{- (j)} - \vartheta_i^{+ (j)}), \quad \text{stoichiometric coefficient}$$

conservation equation for the species

$$\rho D Y_i / Dt = -\nabla \cdot \mathbf{J}'_i + \sum_j \vartheta_i^{(j)} m_i \dot{W}^{(j)},$$

$$\rho \frac{D Y_i}{D t} = \nabla \cdot (\rho D_i \nabla Y_i) + \sum_j \vartheta_i^{(j)} m_i \dot{W}^{(j)} (T, p, \dots Y_k \dots)$$

sum over the reactions

equation for the chemical energy

$$e_{chem} \equiv \sum_i h_i Y_i$$

enthalpy of formation per unit of mass of species i

$$Q^{(j)} = \sum_{i=1}^n (\vartheta_i^{(j)+} - \vartheta_i^{(j)-}) m_i h_i(T_o),$$

sum over the species

heat of the j th reaction

$$\rho D e_{chem} / Dt = -\sum_i \nabla \cdot (h_i \mathbf{J}'_i) - \sum_j Q^{(j)} \dot{W}^{(j)}$$

Thermal balance of **inviscid** flow of **reactive** gas

$$e_{chem} \equiv \sum_i h_i Y_i \quad \rho D e_{chem} / Dt = - \sum_i h_i \nabla \cdot \mathbf{J}'_i - \sum_j Q^{(j)} \dot{W}^j$$

heat released by the j^{th} reaction ↗ ↖ rate of the j^{th} reaction
(number per unit time and unit volume)

$$\rho D(e_T + e_{chem}) / Dt = - \nabla \cdot \mathbf{J}'_q - p \nabla \cdot \mathbf{u}$$

heat flux $\mathbf{J}_q \equiv \mathbf{J}'_q - \sum_i h_i \mathbf{J}'_i \quad \mathbf{J}_q = -\lambda \nabla T$

$$\delta e_T = c_v \delta T \quad c_v \approx \text{cst. (for simplicity, can be easily removed)}$$

$$\rho c_v DT / Dt = \nabla \cdot (\lambda \nabla T) - p \nabla \cdot \mathbf{u} + \sum_j Q^{(j)} \dot{W}^{(j)}$$

continuity $\Rightarrow -p \nabla \cdot \mathbf{u} = \frac{p}{\rho} \frac{D}{Dt} \rho = \frac{D}{Dt} p - \rho \frac{D}{Dt} [(c_p - c_v) T]$

ideal gas law $p = (c_p - c_v) \rho T.$

thermal equation of an inviscid fluid

$$\rho c_p DT / Dt = Dp / Dt + \nabla \cdot (\lambda \nabla T) + \sum_j Q^{(j)} \dot{W}^{(j)}$$

compression conduction chemistry

Governing equations for inviscid flows of reactive gas

$$\frac{1}{\rho} \frac{D\rho}{Dt} = -\nabla \cdot \mathbf{u}, \quad \rho \frac{D\mathbf{u}}{Dt} = -\nabla p, \quad p = (c_p - c_v)\rho T,$$

$$\rho c_p \frac{DT}{Dt} = \frac{Dp}{Dt} + \nabla \cdot (\lambda \nabla T) + \sum_j Q^{(j)} \dot{W}^{(j)},$$

$$\rho \frac{DY_i}{Dt} = \nabla \cdot (\rho D_i \nabla Y_i) + \sum_j \nu_i^{(j)} m_i \dot{W}^{(j)}.$$

stoichiometric coefficient

Conservative form of the energy equation (inviscid approximation)

$$\partial(\rho e_{tot})/\partial t = -\nabla \cdot \mathbf{J}_{e_{tot}} \quad \mathbf{J}_{e_{tot}} = \mathbf{J}'_{e_{tot}} + \rho e_{tot} \mathbf{u} \quad \mathbf{J}_q + \sum_i h_i \mathbf{J}'_i = \mathbf{J}'_{e_{tot}} - p \mathbf{u}$$

$$\partial(\rho e_{tot})/\partial t = -\nabla \cdot [\rho \mathbf{u} e_{tot} + \mathbf{u} p + \mathbf{J}_q + \sum_i h_i \mathbf{J}'_i] = -\nabla \cdot [\rho \mathbf{u} (e_{tot} + p/\rho) + \mathbf{J}_q + \sum_i h_i \mathbf{J}'_i]$$

convective flux of enthalpy

diffusive flux of total energy

progress variable

$$\rho q_m \frac{D\psi}{Dt} \equiv \sum_j Q^{(j)} \dot{W}^{(j)}, \quad \psi \in [0, 1]$$

heat released per unit mass

$$e_{tot} + p/\rho = c_p T + |\mathbf{u}|^2/2 - q_m \psi$$

Viscous flow

$$\underline{\underline{\Pi}} = p\underline{\underline{\mathbf{I}}} + \underline{\underline{\pi}}$$

$$\begin{array}{c} \text{shear viscosity} \swarrow \quad \searrow \text{bulk viscosity} \\ \underline{\underline{\pi}} \equiv -2\eta(\underline{\underline{\nabla}}\underline{\underline{\mathbf{u}}})^{(s)} - \underline{\underline{\mathbf{I}}}(\xi - 2\eta/3)\nabla \cdot \underline{\underline{\mathbf{u}}} \end{array}$$

$$\partial(\rho e_{tot})/\partial t = -\nabla \cdot [\rho \mathbf{u} (c_p T + |\mathbf{u}|^2/2 - q_m \psi) + \mathbf{J}'_q + \mathbf{u} \cdot \underline{\underline{\pi}}]$$

one-dimensional compressible flow

$$\mu \equiv 4\eta/3 + \xi$$

$$\frac{\partial \rho}{\partial t} = -\frac{\partial(\rho u)}{\partial x} \qquad \frac{\partial(\rho u)}{\partial t} = -\frac{\partial}{\partial x} \left(p + \rho u^2 - \mu \frac{\partial u}{\partial x} \right)$$

$$\frac{\partial(\rho e_{tot})}{\partial t} = -\frac{\partial}{\partial x} \left[\rho u (c_p T + u^2/2 - q_m \psi) - \lambda \frac{\partial T}{\partial x} - \mu u \frac{\partial u}{\partial x} \right]$$

(simplest form of heat flux)

II – 6) Entropy production

$$s(\rho, T, ..Y_i, ..)$$

entropy is a function of state that is not a conserved quantity

$$\partial(\rho s)/\partial t = -\nabla \cdot \mathbf{J}_s + \dot{w}_s,$$

2nd law of thermodynamics dissipative effects $\Rightarrow \dot{w}_s \geq 0$

$$T\delta s = \delta e_T + p\delta v - \sum_i \mu_i \delta Y_i$$

ideal gas $\frac{(s - s_o)}{c_v} = \ln \left(\frac{p/\rho^\gamma}{p_o/\rho_o^\gamma} \right)$

$$T \frac{Ds}{Dt} = \frac{De_T}{Dt} + p \frac{D(1/\rho)}{Dt} - \sum_i \mu_i \frac{DY_i}{Dt}$$

$$\dot{w}_s = \mathbf{J}'_q \cdot \nabla \left(\frac{1}{T} \right) - \sum_i \mathbf{J}'_i \cdot \nabla \left(\frac{\mu_i}{T} \right) - \frac{1}{T} \underline{\underline{\pi}} : (\underline{\underline{\nabla u}})^{(s)}$$

inert mixture in one-dimensional geometry

$$J_s = \rho u s - \frac{\lambda}{T} \frac{\partial T}{\partial x},$$

$$\dot{w}_s = \frac{\mu}{T} \left(\frac{\partial u}{\partial x} \right)^2 + \frac{\lambda}{T^2} \left(\frac{\partial T}{\partial x} \right)^2$$

$$\mu > 0, \quad \lambda > 0$$

$$\rho \frac{Ds}{Dt} = \frac{\partial}{\partial x} \left(\frac{\lambda}{T} \frac{\partial T}{\partial x} \right) + \dot{w}_s$$

$$\rho T \frac{Ds}{Dt} = \frac{\partial}{\partial x} \left(\lambda \frac{\partial T}{\partial x} \right) + \mu \left(\frac{\partial u}{\partial x} \right)^2$$